# DYNAMICS OF VOLATILE ORGANIC COMPOUNDS IN MASSON PINE (*PİNUS MASSONİANA* LAMB.) SEEDLING NEEDLES UNDER DROUGHT STRESS

WENXUAN QUAN<sup>1, 2</sup>, CHAO ZHAO<sup>1,2</sup> AND GUIJIE DING<sup>1\*</sup>

<sup>1</sup>Guizhou Provincial Forest Resources and Environment Research Centre / Key Laboratory of Plateau-mountain Forest Cultivation, Guizhou University, Guiyang 550025, China <sup>2</sup>Key Laboratory of Mountainous Environmental Protection, Guizhou Normal University, Guiyang 550001, China \*Corresponding author's email: gjdinggzu@126.com

### Abstract

The aim of this study is to evaluate whether volatile organic compounds (VOCs) of Masson pine adapted their physiological responses under drought stress or not. The needle length, root length, root/shoot ratio and the VOCs of Masson pine seedlings under drought stress were investigated by comprehensive two-dimensional gas chromatography time of flight mass spectrometry (GC×GC-TOFMS). The results showed that the growth index was enhanced with the increase of stress strength, and the seedling growth, biomass, needle length, and root length were inhibited compared with the well-watered treatment. The amounts of total VOCs under drought stress were increased, the concentration of terpenoids, acids and esters, under drought stress were higher than the well-watered treatment, but the concentration of alcohols and carbonyl compounds was slightly decreased under the moderate drought treatment. Light drought, moderate drought, and severe drought compared with the well-watered treatment, increased the release of VOCs by 49, 50, and 49%, respectively. The results show a key role of the VOCs released in the needles and therefore play a possible key role in drought resistance. This study provides evidence that the needles of Masson pine release more VOCs to adapt to drought conditions.

Key words: Masson pine; Seedlings; Drought; VOCs; GC×GC-TOFMS.

**Abbreviations:** VOCs: volatile organic compounds; 2D-GC×GC/TOFMS: comprehensive two-dimensional gas chromatography time of flight mass spectrometry; SWC: soil water content; WW: well-watered; LD: light drought; MD: moderate drought; SD: severe drought; SPME: solid-phase microextraction; The mean  $\pm$  SD: the mean and standard deviation; RSD: relative standard deviation; TIC: total ion chromatogram

# Introduction

The Masson pine (*Pinus massoniana* Lamb.) has been considered one of the pioneer tree species adapted to the Red Earth Area of subtropical China (Zhang *et al.*, 2013). The species is a common tree in plantation forestry that replaces or compensates for the loss of the natural forest, it is a fast-growing tree widely grown in central and southern China (Chau & Lo, 1980; Chen *et al.*, 2015a). Chinese rosin is obtained mainly from the turpentine of this pine, the wood is widely used to make pulp for the paper industry, and the pine needles were used as medicine for a long time (Feng *et al.*, 2010; Yang *et al.*, 2010; Zhang *et al.*, 2011). Because of this, it has become an important tree for afforestation in South China.

Masson pine is the major pioneer species for vegetation reforestation due to its strong adaptability, preferable tolerance to drought and infertility in southern China. However, most of those forests show poor stand structure and low productivity due to human or natural factors, so that the composition of forest ecosystems is reduced, covered with exiguous vegetation and subjected to severe soil erosion, resulting in the poor eco-efficiency of such forests in the conservation of water and soils. Accordingly, the water absorption capacity of the Masson pine root system, as well as mycorrhizal fungi, have been suggested as likely explanations of this adaptation, and Masson pine has become one of the major pioneer species of vegetation reforestation. It has strong resistance to the drought, low phosphorous, acid rain, and the other extreme conditions (Wang et al., 2013; Hu et al., 2014; Li et al., 2017; Zhang et al., 2017).

The biosynthesis and storage' of volatile organic compounds (VOCs), is much more important issue when working with the VOC storing pine needles (Loreto & Schnitzler, 2010). VOCs play an important role in the stress resistance of the coniferous trees (Su *et al.*, 2009; Ioannou *et al.*, 2014; Quan & Ding, 2017). Several recent studies have focused on the chemical nutrient elements and biomedical prospects of Masson pine seedlings (Mohamed *et al.*, 2014; Guan *et al.*, 2015; Wang *et al.*, 2015). Some scientific papers dedicated the comparative proteomic analyses on Masson pine under the acid rain stress (Hu *et al.*, 2014). Therefore, it's necessary to analyze the composition and changes of needle VOCs to provide scientific guidance for the artificial afforestation of Masson pine.

Drought is an environmental constraint that seriously affects the plant growth and productivity, hindering the development of plant production, and therefore monitoring the VOCs changes in Masson pine seedling under drought conditions is necessary. Therefore, the objective of this research aimed to evaluate the growth and needle VOCs responses of Masson pine seedlings to different level of drought stress conditions.

### **Materials and Methods**

**Study site:** The seeds of Masson pine used in this experiment were from the same mother tree in Majiang County, Guizhou Province. Sowing time was in December 2015, and the one-year-old Masson pine seedlings were used for the drought stress in April 2016, which were potted in the greenhouse  $(26^{\circ}35' \text{ N}, 106^{\circ}43' \text{ E})$ .

SWC (g 
$$g^{-1}$$
) = (W1-W2)/W2

where W1 is the fresh weight of soil, and W2 is the dry weight of soil (Watt *et al.*, 1994).

The soil moisture content was maintained by weighing method (Quan & Ding, 2017). The experimental seedlings were divided into 4 groups for processing: well-watered (WW, 70-80% field moisture capacity), light drought (LD, 60-70% field moisture capacity), moderate drought (MD, 50-60% field moisture capacity), and severe drought (SD, 35-50% field moisture capacity) (Xu & Zhou, 2011). The measurements were carried out 60 d after the seedlings were subjected to relatively long-term drought stress treatment, and each group had 12 pots. After treatment, the saplings were used for this study. The needles were harvested and analyzed at the same time. The shoots and roots of the seedlings of each treatment were harvested separately to determine the root/shoot ratio (Chen *et al.*, 2015b).

Pre-treatment method: Seedling needles were cut into pieces, and SPME procedures followed the method with some minor changes (Xiang et al., 2015). The samples (20 mg) were added to the headspace bottle and phenylethyl acetate solution (1 µL) was also added. The headspace bottle was placed in a dry heat block adjusted to 80.0°C for 10 min, and then the SPME fiber (Supelco, PDMS/DVB/CAR) was manually inserted into the sample bottle headspace for 15 min (Xiang et al., 2014). The SPME-coated fiber with trapped VOCs was introduced manually into the GC×GC two-dimensional gas chromatography time of flight mass spectrometry (GC×GC-TOFMS, ZOEX, ZX-1) injection port to allow desorption. The bottle with cap as the headspace bottle (Agilent, 22mL), and phenylethyl acetate as the internal standard (Sigma-Aldrich, purity >99.0%). All measurements were made 3 repetitions per treatment.

**Instrument condition:** The GC×GC-TOFMS were as follows: DB-1MS phase (30 m × 0.25 mm, 1.00  $\mu$ m), second capillary column was DB-wax phase (1.2 m × 0.1 mm, 0.10  $\mu$ m); temperature program: 60°C (held for 2 min) to 240°C (held for 1 min), increased at 3°C min<sup>-1</sup>; inlet pressure: 30 psi (hold 2 min) to 48 psi (hold 1 min), 0.3 psi min<sup>-1</sup>, and the carrier gas was helium; running time: 63 min, inlet temperature: 250°C, injection mode: splitless, purge time: 3.5 min, modulation period: 9 s, continuous period: 300 ms, the hot jet: 375°C and the pressures is 40

psi, the cold jet nitrogen flow rate:18 L min<sup>-1</sup>. The characteristics of the TOFMS system were as follows: ion source: 70 eV EI; ion source temperature: 280°C; transfer line temperature: 300°C; mass range: 45-450 amu; FWHM resolution: 4000; and acquisition rates: 100 Hz.

**Statistical analysis:** The mean and standard deviation (SD) values of three replicates were calculated. Clustering analysis using R to draw a heatmap from VOCs data (Eisen *et al.*, 1998). The data were processed using the Zoex GC-Image system. The qualitative identification of the VOCs of the Masson pine was retrieved by the standard spectral library (NIST2.0 mass spectral library and Wiley08 mass spectral library). The quantitative analysis use the internal standard method (Peng *et al.*, 2004; Xiang *et al.*, 2015), in which phenylethyl acetate served as the internal standard without considering calibration factors.

# Results

Growth status for seedlings of Masson pine under drought stress: The experimental drought stress significantly reduced the needle length, root length and biomass. In general, the seedling biomass for each treatment was affected by the drought stress, except that no significant difference in the seedling height for Masson pine seedlings occurred between the wellwatered and other treatments (Table 1).

Drought stress significantly inhibits the increase of seedling biomass, the aboveground, belowground and total biomass of seedlings under drought stress is lower than those of the well-watered treatment. The root length was the highest under light drought stress (22.10 cm), and the root/shoot ratio was the highest under moderate drought stress. moderate drought stress had been found to promote root/shoot ratio and hence improve the water-use efficiency (Franco *et al.*, 2011; Wurzburger & Miniat, 2014), which may be one of the major reasons that the moderate drought promote the root/shoot ratio.

**The VOCs of Masson pine seedling needle samples:** To confirm the feasibility of the VOCs determination method used in this experiment, the qualitative and quantitative verification of VOCs was also carried out. The chromatogram is shown in Fig. 1.

For the chemometric analysis, the concentrations of 68 VOCs were analyzed (Table 2). The comprehensive analyses of the needle VOCs showed significant quantitative and qualitative differences between the samples. The terpenoids dominated the needle VOCs, with the presence of  $\alpha$ -pinene and caryophyllene characterizing most of the needle VOCs.

Table 1. The Masson pine seedlings status under different drought stress treatments.

Treatments	Needle length (cm)	Root length (cm)	Seedling height (cm)	Shoot dry weight (g)	Root dry weight (g)	Root/shoot ratio
WW	$9.55 \pm 0.88a$	$16.45\pm0.66b$	$19.27 \pm 2.58a$	$1.82\pm0.02a$	$0.59 \pm 0.09a$	$0.32\pm0.05b$
LD	$7.11 \pm 0.76 b$	$22.10 \pm 1.42a$	$18.98 \pm 1.93a$	$1.07\pm0.09b$	$0.44\pm0.04b$	$0.41 \pm 0.02a$
MD	$5.87 \pm 0.72c$	$12.51 \pm 1.67c$	$18.33 \pm 1.60a$	$1.06 \pm 0.11b$	$0.49 \pm 0.06ab$	$0.46 \pm 0.07a$
SD	$5.47 \pm 0.45c$	$9.79\pm0.89d$	$16.87 \pm 1.92a$	$0.99\pm0.04b$	$0.39\pm0.05b$	$0.39 \pm 0.04 ab$

Mean ± standard deviation (n=12). Values sharing same letters differ non-significantly (p>0.05)



Fig. 1. Total ion chromatogram (TIC) of VOCs from the Masson pine needles.

This is an earlier investigation by 2D-GC×GC-TOFMS on the chemical composition of Masson pine needle VOCs under drought stress (Quan & Ding, 2017). In the chemical composition, we classified these VOCs following the metabolic pathways:

- (1) The groups of terpenoids. Terpenoids are dimers, or combinations of a 5 carbon precursor called isoprene, are a large and diverse class. The terpenoids group is a basic VOC substance in pine needles, and it is the main flavor component of pine needles. The terpenoids concentration of LD, MD, and SD are 1.67, 2.03, and 2.58 times of the WW treatment. Terpene hydrocarbons include  $\alpha$ -pinene, camphene,  $\beta$ -thujene, terpinolen,  $\beta$ cubebene, caryophyllene,  $\beta$ -caryophyllene, isoledene,  $\beta$ cadinene, and calamenene. The oxygenated terpenes include borneol,  $\alpha$ -terpieol, isoborneol,  $\beta$ -cyclocitral, bornyl acetate, cedran-8-ol, and manoyl oxide. The norsoprenoids include  $\beta$ -ionone,  $\alpha$ -ionone, dihydroactinidiolide, and geranyl acetone.
- (2) The groups of acids and esters. The acids and esters group is an important group of VOCs in pine needles, and the content of each component has a great influence on Masson pine; the acids and esters group is represented mainly by undecanoic acid, 10-methyl-, methyl ester and benzeneacetic acid, methyl ester. The acids and esters concentration of LD, MD, and SD were 1.51, 1.52, and 3.65 times of the WW treatment, respectively. This group included acetic acid, 2-methyl-2-butenoic acid, hexanoic acid, methyl ester, (E)-3-hexenoic acid, methyl ester, (E)-(E)-2-hexenoic 2,4-3-hexenoic acid, acid. hexadienoic acid, methyl ester, methyl (3E)-3nonenoate, butanedioic acid, methyl-, dimethyl ester,

benzoic acid, methyl ester, octanoic acid, methylester, benzoic acid, benzeneacetic acid, methyl ester, nonanoic acid, methyl ester, acetic acid, 2phenylethyl ester, decanoic acid, methyl ester, nonanoic acid 9-oxo-methyl ester, undecanoic acid 10-methyl-methyl ester, (Z)-hex-3-enyl benzoate, methyl tetradecanoate, hexadecanoic acid, methyl ester, and thymol methyl ether.

- (3) The groups of alcohols. The alcohols group is an important VOC substance in pine needles, and represented mainly by 1-octanol. The alcohols concentration of LD, MD, and SD were 1.25, 0.70, and 2.67 times of the WW treatment. The alcohols included 1-penten-3-ol, 2-(Z)-pentenol, 3-(Z)-hexen-1-ol, 1-hexanol, 1-nonen-3-ol, benzenemethanol, 2-ethyl-1-hexanol, 1-octanol, benzeneethanol, and 6-methyl-5-hepten-2-ol.
- (4) The groups of carbonyl compounds. The carbonyl compounds group is an important VOCs substance in pine needles, they are emitted from a variety of plants, and they affect plant growth. Most parts of carbonyl compounds can be metabolized into organic acids, amino acid, etc., by the endoenzymes in needles. The carbonyl compounds concentration of LD, MD, and SD were 1.17, 0.76, and 1.45 times of the WW treatment. The compounds included 1-penten-3-one, 2-methyl butanal, 2-(E)-pentenal, 2-methyl-4-pentenal, hexanal, 2-hexenal, (E,E)-2,4-hexadienal, benzaldehyde, 5-hepten-2-one, 6-methyl-, octanal, benzeneacetaldehyde, and decanal.
- (5) The others. The other components were detected in lower percentages in needles, not exceeding 1%. It includes 2-ethylfuran, pyridine, and 5-ethyl-2(5H)-furanone.

Table 2.	<b>Oualitative and</b>	quantitative anal	vsis of vola	tile components i	in needles of	Masson pine seedlings.
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	<b>Example 2.</b> Quantative and quantitative analysis of volatile components in needles of wildsson pine seedings. <b>Polotive cone + SD</b> $(ng e^{-1})$							
ID	Compound name	Formula	WW	LD	MD	SD		
1	Acetic acid	C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	66 82 + 4 73	191 56 ± 13 65	209 11 + 11 37	368.06 + 25.47		
2	1-Penten-3-one	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	$41.89 \pm 2.56$	$216.16 \pm 15.03$	41.12 + 4.15	48.22 + 4.11		
3.	1-Penten-3-ol	C <sub>5</sub> H <sub>10</sub> O	$61.18 \pm 3.36$	158.41 + 9.98	$66.86 \pm 4.55$	126.06 + 10.72		
4.	2-Methyl butanal	$C_5H_{10}O$	$59.54 \pm 3.28$	$42.77 \pm 3.69$	$67.81 \pm 5.61$	$21.13 \pm 1.91$		
5.	2-Ethylfuran	C <sub>6</sub> H <sub>8</sub> O	$18.45 \pm 1.15$	$14.31 \pm 0.91$	$11.58 \pm 1.79$	$12.88 \pm 1.38$		
6.	Pyridine	C <sub>5</sub> H <sub>5</sub> N	$153.23\pm8.43$	$34.07 \pm 2.15$	$118.20\pm9.04$	$116.83 \pm 10.93$		
7.	2-(E)-Pentenal	C <sub>5</sub> H <sub>8</sub> O	$76.21 \pm 4.29$	$184.58 \pm 13.63$	$83.96 \pm 6.71$	$177.06\pm8.05$		
8.	2-(Z)-Pentenol	$C_5H_{10}O$	$35.40 \pm 1.97$	$340.88\pm31.48$	$29.69\pm3.02$	$33.32\pm3.83$		
9.	2-Methyl-4-pentenal	$C_6H_{10}O$	$213.44\pm15.74$	$235.10\pm24.81$	$159.57 \pm 16.85$	$222.83\pm15.94$		
10.	Hexanal	$C_6H_{12}O$	$317.90\pm27.48$	$240.51 \pm 25.15$	$166.58 \pm 13.13$	$252.66\pm18.48$		
11.	2-Hexenal	$C_6H_{10}O$	$3834.42 \pm 326.23$	$5630.97 \pm 454.75$	$3513.49 \pm 338.92$	$3760.46 \pm 319.64$		
12.	3- (Z)-Hexen-1-ol	$C_6H_{12}O$	$5764.04 \pm 441.58$	$8855.35 \pm 657.89$	$2664.99 \pm 282.22$	$7154.81 \pm 608.16$		
13.	1-Hexanol	$C_6H_{14}O$	$743.29 \pm 43.85$	$715.36 \pm 65.07$	$308.97 \pm 31.01$	$1366.82 \pm 116.18$		
14.	(E,E)-2,4-Hexadienal	$C_6H_8O$	$239.13 \pm 14.11$	$399.47 \pm 26.17$	$169.60 \pm 12.53$	$189.47 \pm 17.10$		
15.	2-Methyl-2-Butenoic acid	$C_5H_8O_2$	$117.01 \pm 6.92$	$51.73 \pm 4.26$	$30.87 \pm 3.15$	$457.92 \pm 48.71$		
16.	Hexanoic acid, methyl ester	$C_7H_{14}O_2$	$135.44 \pm 7.99$	$120.72 \pm 7.61$	$172.08 \pm 12.70$	279.69 ± 23.77		
17.	(E)-3-Hexenoic acid, methyl ester	$C_7H_{12}O_2$	3446.98 ± 313.37	$2790.74 \pm 275.82$	$4012.58 \pm 372.86$	$2494.92 \pm 184.60$		
18.	5-ethyl-2(5H)-Furanone	$C_6H_8O_2$	$31.35 \pm 1.87$	$54.71 \pm 4.45$	$30.85 \pm 3.10$	$28.17 \pm 3.30$		
19.	Benzaldehyde	$C_7H_6O$	$2084.37 \pm 122.98$	$1246.48 \pm 88.53$	$824.23 \pm 66.05$	$5218.16 \pm 407.02$		
20.	a-Pinene	$C_{10}H_{16}$	8289.26 ± 589.07	$10323.38 \pm 850.37$	$13924.24 \pm 846.85$	$18502.22 \pm 1143.17$		
21.	(E) 2 Heyeneis asid	$C_{10}H_{16}$	$51.85 \pm 5.10$	$44.09 \pm 3.82$	$105.94 \pm 7.21$	$140.08 \pm 10.93$		
22.	(E)-5-nexelloic acid	$C_{6}\Pi_{10}O_{2}$	$964.40 \pm 39.16$ $106.20 \pm 7.27$	$221.11 \pm 25.93$ 168.02 ± 11.64	$228.89 \pm 13.30$ $265.02 \pm 10.08$	$232.02 \pm 16.00$ 100.25 ± 16.15		
23. 24	1 Nopen 2 ol	$C_8\Pi_{14}O$	$100.20 \pm 7.27$	$108.93 \pm 11.04$	$203.93 \pm 19.08$ $04.20 \pm 0.41$	$199.33 \pm 10.13$ $170.00 \pm 14.78$		
24. 25	(F)-2-Hevenoic acid	$C_{9}\Pi_{18}O$	$39.49 \pm 3.43$ $11.95 \pm 2.67$	$15.00 \pm 3.10$	$94.30 \pm 9.41$ 16 97 + 1 45	$170.09 \pm 14.78$ 33.27 + 2.59		
25. 26	2 4-Hexadienoic acid methyl ester	$C_{6}H_{10}O_{2}$	$30553 \pm 1813$	$13.99 \pm 1.21$ 271 98 + 18 13	348.28 + 31.68	47759 + 3768		
20.	Octanal	$C_{H_0O_2}$	44 11 + 272	$34.07 \pm 2.16$	5357 + 364	$61.93 \pm 5.02$		
28	Methyl (3E)-3-nonenoate	$C_{10}H_{18}O_2$	$50.56 \pm 2.98$	$57.77 \pm 3.64$	$28.52 \pm 1.34$	$32.24 \pm 2.71$		
29.	Benzenemethanol	C7H8O	$212.52 \pm 12.54$	$180.33 \pm 11.46$	$95.85 \pm 4.52$	$214.15 \pm 18.35$		
30.	2-Ethyl-1-hexanol	$C_8H_{18}O$	$74.96 \pm 4.12$	$46.56 \pm 3.93$	$20.82 \pm 1.42$	$140.50 \pm 12.38$		
31.	Benzeneacetaldehyde	C <sub>8</sub> H <sub>8</sub> O	$134.45 \pm 8.87$	$64.78 \pm 4.18$	$76.05 \pm 5.30$	$129.68 \pm 11.50$		
32.	Butanedioic acid, methyl-, dimethyl ester	$C_7H_{12}O_4$	$493.29 \pm 31.08$	$230.97 \pm 14.55$	$260.39 \pm 15.49$	$745.14 \pm 60.36$		
33.	$\beta$ -Thujene	$C_{10}H_{16}$	$186.71 \pm 15.76$	199.65 ± 13.58	$668.75 \pm 47.48$	$782.28 \pm 58.35$		
34.	1-Octanol	$C_8H_{18}O$	$2746.70 \pm 193.04$	$1731.25 \pm 119.07$	$3092.07 \pm 289.54$	$15238.86 \pm 1676.27$		
35.	Benzoic acid, methyl ester	$C_8H_8O_2$	$137.18\pm9.64$	$175.73 \pm 12.07$	$169.11 \pm 15.01$	$738.54 \pm 81.24$		
36.	Terpinolen	$C_{10}H_{16}$	$883.65 \pm 75.67$	$5184.10 \pm 426.61$	$8229.14 \pm 587.27$	$11018.93 \pm 910.81$		
37.	Benzeneethanol	$C_8H_{10}O$	$21.24\pm2.17$	$23.82 \pm 1.51$	$39.81 \pm 3.83$	$528.66\pm58.15$		
38.	Octanoic acid, methylester	$C_9H_{18}O_2$	$855.51 \pm 67.05$	$1047.14 \pm 65.97$	$1306.10 \pm 72.73$	$1352.33 \pm 128.76$		
39.	6-Methyl-5-hepten-2-ol	$C_8H_{16}O$	$619.46 \pm 54.07$	$803.42 \pm 51.62$	$840.91 \pm 68.70$	$2608.63 \pm 276.94$		
40.	Benzoic acid	$C_7H_6O_2$	$89.61 \pm 9.93$	$53.33 \pm 3.46$	$64.80 \pm 5.60$	$109.97 \pm 12.10$		
41.	Benzeneacetic acid, methyl ester	$C_9H_{10}O_2$	$2540.45 \pm 239.72$	$4616.18 \pm 391.82$	$4971.10 \pm 352.96$	$13079.36 \pm 1238.73$		
42.	Borneol	$C_{10}H_{18}O$	$571.82 \pm 31.45$	$2120.96 \pm 143.62$	$1182.08 \pm 79.93$	$4279.69 \pm 410.77$		
43.	Decanal	$C_{10}H_{20}O$	87.74 ± 4.83	$40.19 \pm 3.53$	$48.67 \pm 4.46$	$223.46 \pm 20.56$		
44.	a-Terpieol	$C_{10}H_{18}O$	$299.84 \pm 16.69$	$315.50 \pm 29.88$	$505.75 \pm 45.91$	$398.68 \pm 36.68$		
45.	Isoborneol	$C_{10}H_{18}O$	$10.05 \pm 0.78$	$79.13 \pm 5.89$	$50.96 \pm 4.62$	$160.63 \pm 14.58$		
46.	Nonanoic acid, methyl ester	$C_{10}H_{20}O_2$	$18.48 \pm 1.54$	$33.05 \pm 2.18$	$20.15 \pm 2.43$	//.96 ± /.35		
47.	β-Cyclocitral	$C_{10}H_{16}O$	$108.85 \pm 9.49$	$65.63 \pm 5.13$	$(1.5) \pm 6.08$	$358.13 \pm 32.96$		
48.	A setie said 2 mbanylethyl seter	$C_{11}H_{16}O$	$111./8 \pm 8.42$	$84.25 \pm 0.31$	$03.78 \pm 3.53$	$77.74 \pm 7.15$		
49. 50	Acetic acid, 2-phenyletinyl ester	$C_{10}\Pi_{12}O_2$	$76.46 \pm 0.12$ $613.44 \pm 47.05$	$19.00 \pm 2.21$ $322.11 \pm 21.20$	$445.48 \pm 41.63$	$30.45 \pm 3.19$ 1120 20 $\pm$ 113 80		
51	Decanoic acid methyl ester	$C_{12}\Pi_{20}O_2$	$013.44 \pm 47.93$ $3/8 92 \pm 27.62$	$322.11 \pm 21.29$ $360.07 \pm 24.68$	$445.46 \pm 41.03$ $455.56 \pm 32.34$	$1129.20 \pm 113.89$ $1702.02 \pm 156.59$		
52	Nonanoic acid, 9-oxo-methyl ester	$C_{11}\Pi_{22}O_2$	$84.02 \pm 27.02$	$255.25 \pm 26.08$	$126.94 \pm 9.01$	$1702.02 \pm 130.37$ $124.42 \pm 11.45$		
53	$\beta$ -Cubebene	CurHa	322.61 + 35.16	$960.32 \pm 20.00$	$120.94 \pm 9.01$ 1950 54 + 88 49	$124.42 \pm 11.43$ 1663 75 + 143 07		
53. 54	<i>a</i> -Jonone	C15H24	$198.82 \pm 15.10$	11649 + 834	$111 47 \pm 50.47$	$103.89 \pm 6.03$		
55	Geranyl acetone	$C_{13}H_{20}O$	$99.88 \pm 4.79$	$90.51 \pm 6.70$	10972 + 779	$84.02 \pm 0.05$		
56.	Carvophyllene	C15H220	7174.61 + 659.62	12246.83 + 971.55	13791.52 + 979.20	$18405.85 \pm 1067.54$		
57.	Manovl oxide	$C_{20}H_{34}O$	$3187.49 \pm 348.62$	$3504.26 \pm 421.77$	$31.24 \pm 3.22$	$11.54 \pm 0.77$		
58.	β-Ionone	$C_{13}H_{20}O$	$230.76 \pm 18.45$	$207.43 \pm 23.07$	$148.40 \pm 11.54$	$267.86 \pm 15.64$		
59.	$\beta$ -Caryophyllene	C15H24	$2770.68 \pm 162.39$	$5700.11 \pm 459.11$	$6894.36 \pm 489.50$	$6162.68 \pm 357.44$		
60.	Undecanoic acid, 10-methyl-methyl ester	$C_{13}H_{26}O_2$	$1305.25 \pm 72.79$	$6151.26 \pm 387.53$	4783.23 ± 339.61	$15504.62 \pm 898.27$		
61.	Isoledene	C <sub>15</sub> H <sub>24</sub>	$323.85 \pm 17.91$	$689.27 \pm 44.42$	$1041.68 \pm 83.96$	$1358.72 \pm 78.81$		
62.	Dihydroactinidiolide	$C_{11}H_{16}O_2$	$44.06\pm2.42$	$29.27 \pm 1.84$	$24.53 \pm 2.74$	$67.81 \pm 5.93$		
63.	β-Cadinene	C15H24	$383.38\pm21.79$	$867.20\pm54.63$	$2134.13 \pm 131.52$	$838.28\pm60.03$		
64.	Calamenene	$C_{15}H_{24}$	$448.37\pm34.66$	$735.83\pm48.36$	$1701.03 \pm 120.77$	$1974.05 \pm 146.08$		
65.	(Z)-Hex-3-enyl benzoate	$C_{13}H_{16}O_2$	$135.69\pm7.46$	$550.20\pm24.66$	$155.38\pm10.13$	$1097.38 \pm 91.21$		
66.	Cedran-8-ol	$C_{15}H_{26}O$	$71.19\pm5.92$	$83.53\pm6.37$	$84.84 \pm 9.02$	$92.31\pm8.83$		
67.	Methyl tetradecanoate	$C_{15}H_{30}O_2$	$192.56 \pm 17.69$	$371.56\pm23.41$	$206.18\pm9.64$	$2808.51 \pm 207.83$		
68.	Hexadecanoic acid, methyl ester	$C_{17}H_{34}O_2$	$202.58 \pm 11.24$	$130.28\pm8.21$	$135.44 \pm 7.62$	$1026.70 \pm 95.98$		

Mean  $\pm$  standard deviation (n=3)

These VOCs include terpenoids, acids and esters, alcohols, carbonyl compounds, etc. Almost every sample of a different treatment was characterized by the abundance of the group of terpenoids, acids and esters, alcohols. a-pinene, as well as caryophyllene and terpinolen, were the dominant members of the terpenoids group. Drought stress had a profound impact on the concentration of the terpenoids, acids and esters, which were significantly higher under LD, MD and SD than in WW. However, the concentration of alcohols and carbonyl compounds were slightly decreased under the MD treatment, compared with the WW treatment. Using Ri386 to draw a heatmap from VOCs data (Fig. 2). In the heatmap, we could see that the change of terpenoids was the most obvious under four drought treatments, followed by the groups of acids and esters.



Fig. 2. Clustered image maps of the VOCs of needles in the different treatments. The pine needle VOCs are shown in rows and drought treatments are shown in columns. (A. terpenes, B. acids and esters, C. alcohols, D. carbonyl compounds, E. others).

For LD, MD, and SD compared with WW treatment, the amount of VOCs was increased by 49%, 50%, and 49%, respectively. Some main VOCs, such as 3-(Z)-hexen-1-ol, benzaldehyde,  $\alpha$ -pinene, 1-octanol, 6-methyl-5hepten-2-ol, benzeneacetic acid, methyl ester, borneol, decanoic acid, methyl ester,  $\beta$ -cubebene, caryophyllene,  $\beta$ caryophyllene, undecanoic acid 10-methyl-methyl ester, isoledene, calamenene, and methyl tetradecanoate were increased with the deterioration of drought stress. Moreover, only a small number of VOCs were reduced by the drought is deepened, such as (E)-3-hexenoic acid, methyl ester,  $\alpha$ -ionone, and manoyl oxide. The results showed a key role of the VOCs released in the needles and therefore played a possible key role in drought resistance.

#### Discussion

Effects of drought stress on plant growth and biomass: Drought disaster is an important factor that restricts the sustainable development of forestry in south China, and drought has devastating effects on many needle trees (Kolb & Robberecht, 1996; Gonzalez-Moro *et al.*, 1999; Gruber *et al.*, 2010; Swidrak *et al.*, 2011; Vacchiano *et al.*, 2012), and impacts ecosystems and species richness (Klos *et al.*, 2009; Allen *et al.*, 2010; Holmstrup *et al.*, 2012; Weed *et al.*, 2013; Vose *et al.*, 2016). However, a recent study implied that moderate drought enhanced the competitive ability of forest trees (Wurzburger & Miniat, 2014).

The drought also induces changes in the cellular biochemistry of seedlings, such as reduced photosynthesis, a reduced growth rate, altered water balance (Pichler & Oberhuber, 2007; Thabeet *et al.*, 2009). *Pinus* is a staple genus of tree of high economic value in middle and south China. However, drought stress is a severe factor limiting its growth and reducing its biomass. Some previous studies showed that Scots pine was more affected by summer droughts (Merlin *et al.*, 2015). Our results showed that the severe drought condition inhibited the growth of needle and roots.

Effects of drought stress on plant VOCs: The accretion in the concentration of VOCs in Masson pine needles was due to drought stress, as indicated by the unaltered needle length. However, the functioning of the needle tissues did not appear to be altered, as the number of VOCs was not reduced due to the effect of drought stress. When the coniferous trees suffer from the stress, they release VOCs to enhance their defenses, such as Pinus ponderosa (Schade & Goldstein, 2001), Pinus halepensis (Filella et al., 2009), Pinus pinaster (Sampedro et al., 2011; Blanch et al., 2012), and Pinus sylvestris (Achotegui-Castells et al., 2013; Lundborg et al., 2016). The VOCs play a key role in plant resistance (Ruther & Kleier, 2005; Kännaste et al., 2009; Su et al., 2009; Dicke & Loreto, 2010; Loreto & Schnitzler, 2010; Bracho-Nunez et al., 2011; Azeem et al., 2013), and moreover, also a possible key role in pollination (Peñuelas et al., 2014). The efficacy of terpenoids as the research genetic markers, geographic variation and evolution has been resolved (Hanover, 1992). However, some researchers have implied that terpenoids are frequently used as chemosystematic markers due to the genetic control of their biosynthesis (Baradat & Yazdani, 1988; Incerti et al., 2013).

In these VOCs, terpenoids in the Masson pine play a major defensive role (Achotegui-Castells *et al.*, 2013), and  $\alpha$ -pinene is the most abundant metabolites, which is commonly found in the needles (Yatagai & Hong, 1997; Ioannou *et al.*, 2014). In this study, the main VOCs were  $\alpha$ -pinene (12.42%), caryophyllene (12.36%) and terpinolen (7.40%), followed by  $\beta$ -caryophyllene (4.14%) and borneol (2.87%).

The advantage of 2D-GC×GC-TOFMS in investigating the plant VOCs: Generally, GC×GC can identify the large range of target analytes; in addition, GC×GC has the higher resolution and sensitivity than one-dimensional gas chromatography and yields structured chromatograms that facilitate the identification of unknown compounds (Hoggard *et al.*, 2009). The method resolves the overlapping peaks and interference peak and provides useful identification information (Fig. 3).



Fig. 3. Surface views showing the use of 2D in the separation of four VOCs, 2-hexenal, 3- (Z)-hexen-1-ol, benzaldehyde and  $\alpha$ -pinene (A. The peaks (11 and 12) were inseparable and effectively eliminate the interference peak; B. the peaks (19 and 20) were inseparable under the chromatographic conditions used).

In prior studies, solvent extraction and distillation were generally used for extraction of VOCs in Masson pine, and identification by conventional methods based on onedimensional gas chromatography (Su et al., 2009; Liu et al., 2015). At present, only a handful number of studies reported using HS-SPME combined with GC×GC-TOFMS technology, these included tobacco (Zhu et al., 2005; Ding et al., 2013; Xiang et al., 2015), wine (Danielle et al., 2005; Zhu et al., 2007), medicine (Qiu et al., 2007), milk (Yue et al., 2015), honey (Čajka et al., 2007), coffee beans (Humston et al., 2009), toxic waste (Vos et al., 2011), and essential oils (Jiang et al., 2015). The 2D-GC×GC-TOFMS method could be an effective technical means for the identification of VOCs in Masson pine needles, this study provided an insight into changes in the VOCs of coniferous trees as they adapted to drought stress.

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